

# The Environmental Fate of Engineered Nanoparticles in Lake Geneva Water

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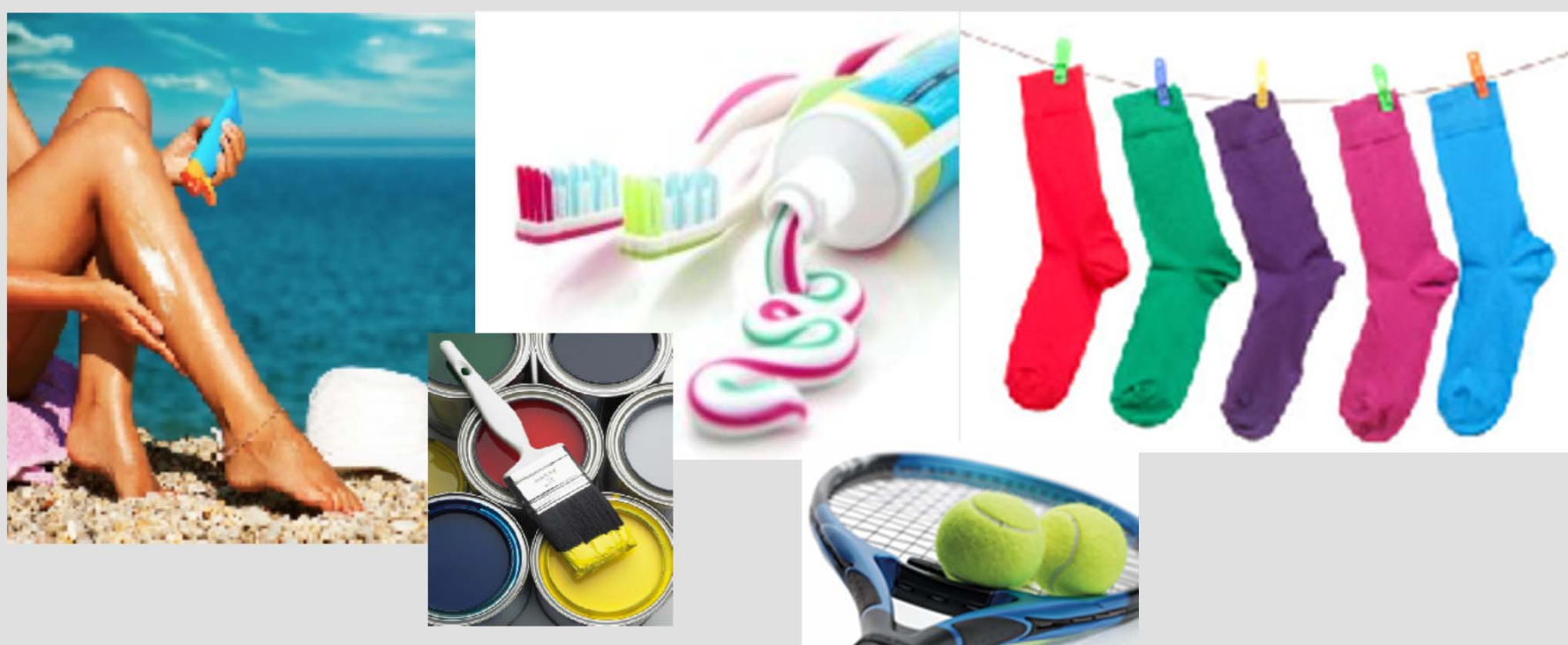


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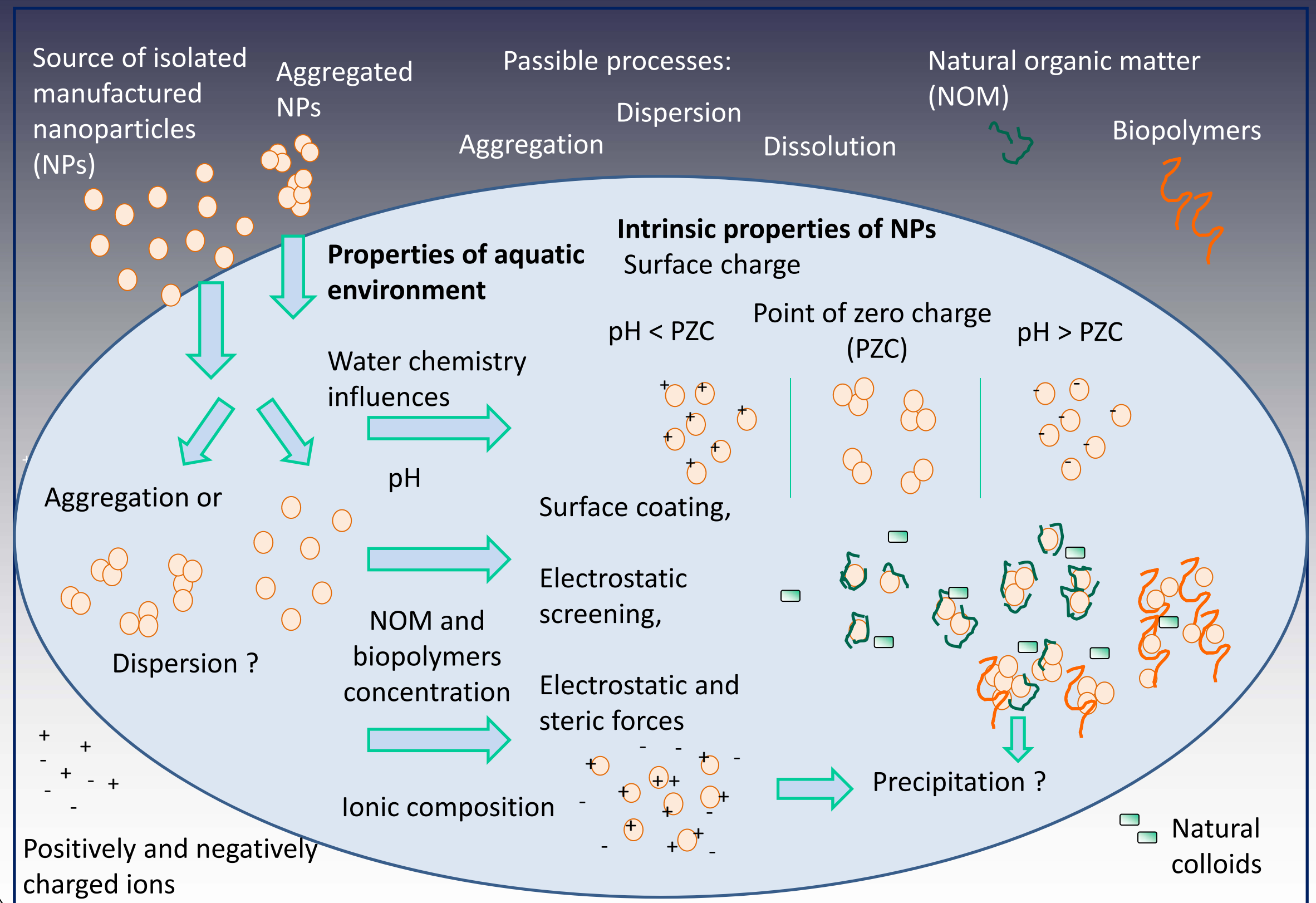
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## Introduction

Cerium dioxide nanoparticles ( $\text{CeO}_2$  NPs) are widely used in many consumer products due to their catalytic and oxidative properties. The growing consumption of such products is constantly increasing the presence of  $\text{CeO}_2$  NPs in the environment, including natural water sources such as rivers or lakes. Despite the number of researches which have been conducted in this area, the fate of  $\text{CeO}_2$  NPs in aquatic systems isn't clearly understood



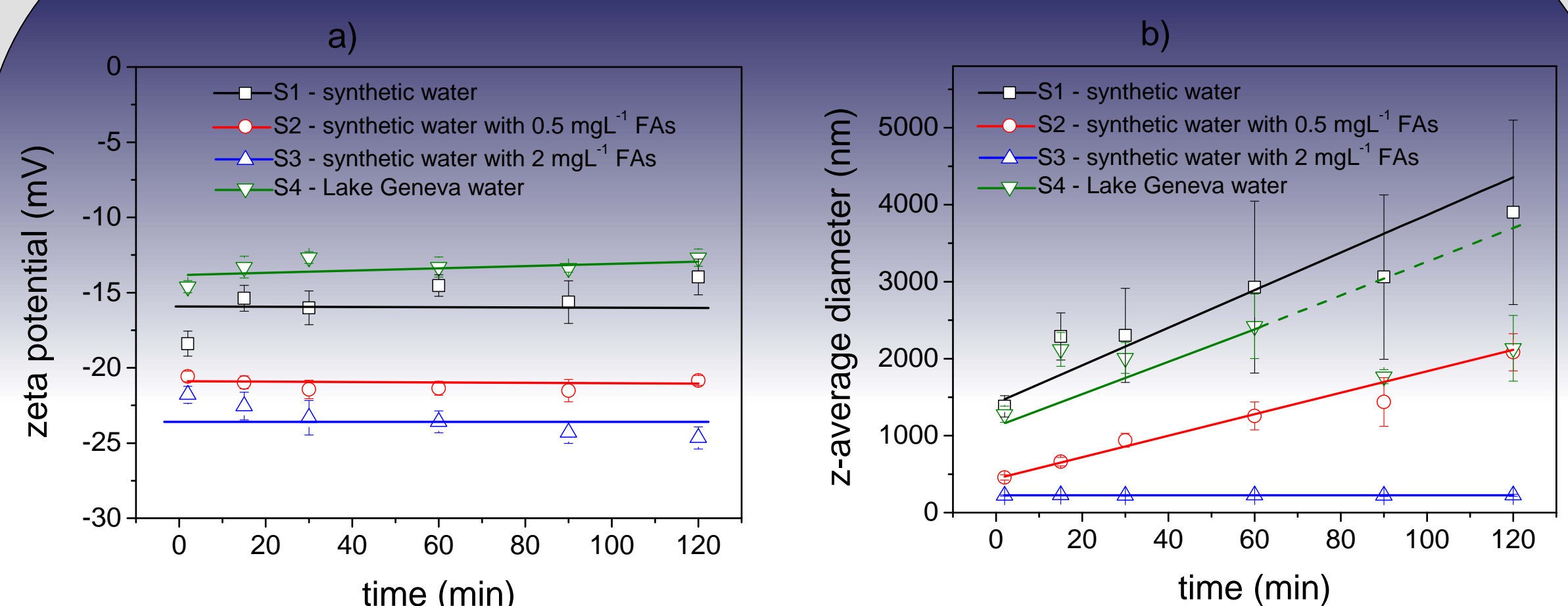
## Nanoparticles fate and behavior in natural fresh water



## Materials and Methods

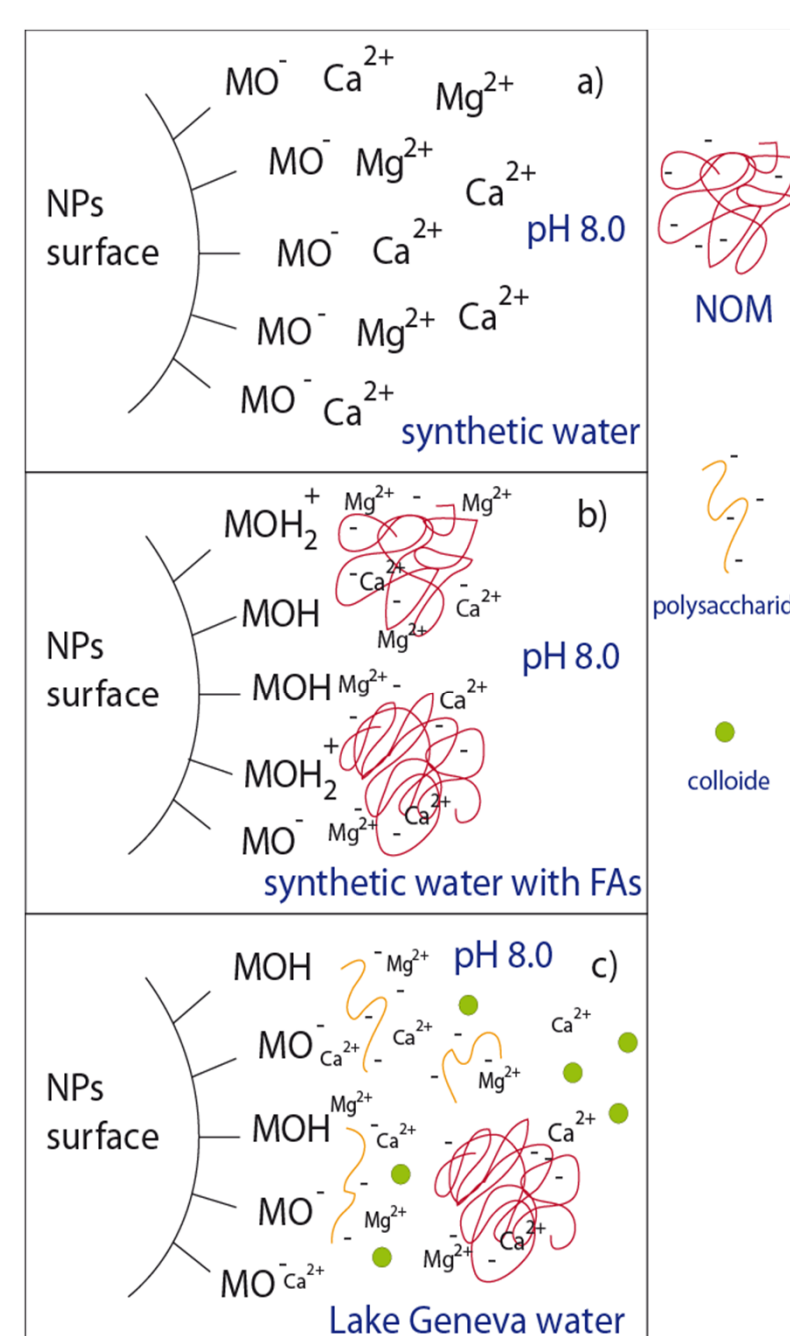
- **$\text{CeO}_2$  NPs (NM-212)** Uncoated from JRC repository (NanoMILE project), Average particle size :  $28.4 \pm 10.4$  nm (manufacturer), Specific Area :  $27.8 \pm 1.5$   $\text{m}^2/\text{g}$  (BET)
- **FAs** Type: Suwannee River Standard II (2S101F) purchased from International Humic Substance Society, USA
- **DLS:** A Malvern Zetasizer Nano ZS was used to determine the zeta potential and z-average hydrodynamic diameter of the  $\text{CeO}_2$  NPs
- **SEM:** A JEOL JSM-7001FA was used for image analysis

## Stability of $\text{CeO}_2$ NPs in synthetic and natural waters



(a) Variation of zeta potentials and (b) z-average hydrodynamic diameters with time in different water samples

## Effect of NOM, natural colloids and ionic strength on stability of $\text{CeO}_2$ NPs



**SEM images** of  $\text{CeO}_2$  NPs in different water samples:

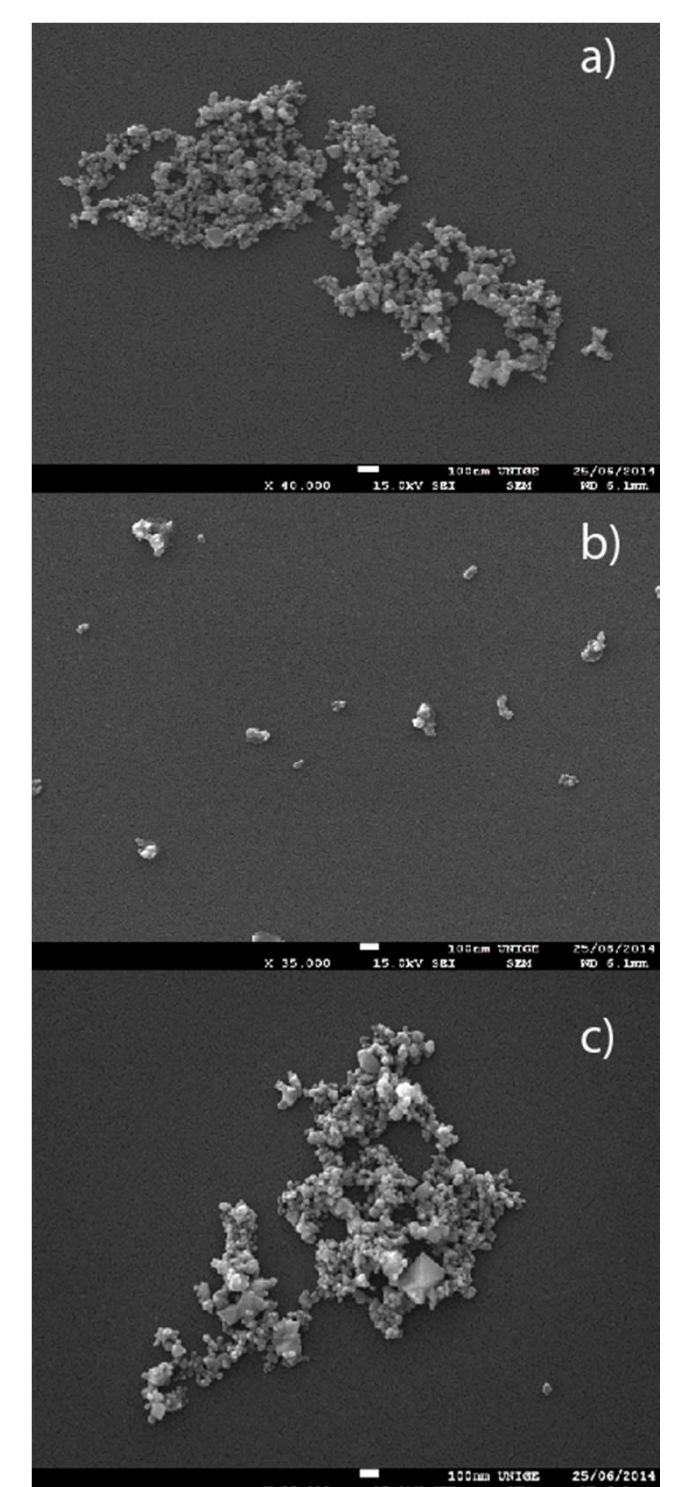
a) synthetic water;

b) synthetic water with  $2 \text{ mgL}^{-1}$  FAs;

In the presence of FAs NPs are stabilized and the size of aggregates is equal to 200 – 250 nm

c) natural Lake Geneva water

In synthetic water and in natural Lake Geneva water NPs are found aggregated and implemented in a matrix made from natural organic matter and salts



Schematic representation of the processes occurring when NPs are released in various water samples

## Conclusions

- pH, ionic strength and coating with natural organic matter play an essential role on the  $\text{CeO}_2$  NPs behavior in the aquatic environment
- The presence of excess of FAs leads to the kinetic stability of  $\text{CeO}_2$  NPs in model systems as ultrapure and synthetic water
- Presence of natural colloids, such as natural organic matters, polyelectrolytes, polysaccharides, biopolymers, iron oxy-hydroxide and inorganic particles in natural Geneva Lake water results in  $\text{CeO}_2$  NPs aggregation

## Reference

Oriekhova, O., Stoll, S., 2016. Effects of pH and fulvic acids concentration on the stability of fulvic acids – cerium (IV) oxide nanoparticle complexes. Chemosphere 144, 131–137. doi:10.1016/j.chemosphere.2015.08.057